

Comment on problem #3. Note that I present the result where λ was not also expanded as a Taylor series. As discussed in class, one needs to do this to get the correct sign for the correction to the ideal gas pressure.

HW #4 Answers.

1. Show that the probability distribution p_i that maximizes the entropy for die rolls subject to a constant value of the 2nd moment is a gaussian.

$$S = -k \sum p_i \ln p_i$$

$$\text{constraints } \sum_{i=1}^6 p_i = 1, \quad \sum_{i=1}^6 i^2 p_i = \text{const}$$

$$\frac{\partial S}{\partial p_i} - \alpha - \delta i^2 = 0 \quad \forall i$$

k has been incorporated into α and δ .

$$-1 - \ln p_i - \alpha - \delta i^2 = 0$$

$$\ln p_i = -1 - \alpha - \delta i^2 \rightarrow p_i = e^{(-1 - \alpha - \delta i^2)} = a e^{-\delta i^2}$$

$$a \sum_{i=1}^6 e^{-\delta i^2} = 1 \Rightarrow p_i = \frac{e^{-\delta i^2}}{\sum e^{-\delta i^2}}$$

n	0	1	2	3	4
f_n / f_0	1.0	0.2	0.04	0.008	0.002

$$f_n = \frac{e^{-\beta h \nu (n + \frac{1}{2})}}{q}, \quad q = \frac{e^{-\beta h \nu / 2}}{(1 - e^{-\beta h \nu})}$$

$$f_1 / f_0 = e^{-\beta h \nu}$$

$$f_2 / f_0 = e^{-2\beta h \nu}$$

$$f_3 / f_0 = e^{-3\beta h \nu}$$

$$f_4 / f_0 = e^{-4\beta h \nu}$$

$$e^{-\beta h \nu} = 0.2 \Rightarrow \beta h \nu = \ln 5 = 1.602$$

$$e^{-2\beta h \nu} = 0.04 \Rightarrow 2\beta h \nu = \ln 25 \Rightarrow \beta h \nu = 1.602, \text{ etc.}$$

Since there is a constant ratio between successive population values, the distribution is at thermal distribution.

What is the temperature, if the molecule is N_2 ?

$$\Theta_v = \frac{h\nu}{k} = 3374 \text{ K}$$

$$\beta h\nu = \frac{h\nu}{kT} = 1.602 \Rightarrow T = 2089 \text{ K}$$

3. Problem 4-9 from McQuarrie

The problem asked one to show that

$pV \geq \langle N \rangle kT$ for fermions and $pV \leq \langle N \rangle kT$ for bosons

$$\beta pV = \pm \sum_j \ln[1 \pm \lambda e^{-\beta \epsilon_j}]$$

Develop this in a Taylor series keeping the first two terms. [Note the text considers the case where only the leading term is kept.]

$$\ln(1+x) \sim x - \frac{x^2}{2}; \quad \ln(1-x) \sim -x - \frac{x^2}{2}$$

$$\text{fermions: } pV = kT \sum_j \left[\lambda e^{-\beta \epsilon_j} - \frac{1}{2} (\lambda e^{-\beta \epsilon_j})^2 \right]$$

$$pV = NkT - \underbrace{\frac{kT}{2} \sum_j \lambda^2 e^{-2\beta \epsilon_j}}_{+ \text{ quantity.}}$$

We have actually shown the opposite of what was stated in the text.

$$\text{bosons: } pV = kT \sum_j \left[\lambda e^{-\beta \epsilon_j} + \frac{1}{2} (\lambda e^{-\beta \epsilon_j})^2 \right]$$

$$pV = NkT + \frac{kT}{2} \sum_j \lambda^2 e^{-2\beta \epsilon_j}$$

Again, this is opposite from what we were asked to prove.

4. A thermodynamic system with three states has $P_1 = 0.9$, $P_2 = 0.09$, and $P_3 = 0.01$ at $T = 300\text{K}$. What are the energies of states 2 and 3 relative to the ground state (1)?

Take the ground state to be the zero of energy

$$P_1 = \frac{1}{1 + e^{-\beta \epsilon_2} + e^{-\beta \epsilon_3}} = 0.9, \quad q = 1/0.9$$

$$P_2 = 0.09 = 0.9 e^{-\beta \epsilon_2} \quad , \quad P_3 = 0.01 = 0.9 e^{-\beta \epsilon_3}$$

$$0.1 = e^{-\beta \epsilon_2} \quad , \quad \beta \epsilon_2 = 2.30$$

$$\epsilon_2 = 2.30 (300\text{K}) (1.38 \times 10^{-23} \text{J/K}) = 9.52 \times 10^{-21} \text{J}$$

$$\text{Similarly} \quad \epsilon_3 = 4.5 (300\text{K}) (1.38 \times 10^{-23} \text{J/K}) = 1.86 \times 10^{-20} \text{J}$$