## **Chapter 7: classical Statistical Mechanics**

So far, we have employed QM and considered the high *T* classical limits

Suppose we assume from the beginning that we can describe the system classically?

We conjecture that 
$$q_{cl} \sim \int \cdots \int e^{-\beta H(p,q)} dp dq$$

monatomic ideal gas – one atom

$$H = \frac{1}{2m} (p_x^2 + p_y^2 + p_z^2)$$

$$q_{tr} = \int \cdots \int e^{\frac{-\beta(p_x^2 + p_y^2 + p_z^2)}{2m}} dp_x dp_y dp_z \underline{dxdydz}$$

$$q_{tr}^{cl} \sim V \left[ \left( \int_{\infty}^{\infty} e^{-\beta p^{2}/2m} dp \right) \right]^{3}$$
$$= \left( 2\pi mkT \right)^{3/2} V$$

but

$$q_{tr}^{quant} = \left(\frac{2\pi mkT}{h^2}\right)^{3/2} V$$

$$H_{rot} = \frac{1}{2I} \left( p_{\theta}^2 + \frac{p_{\phi}^2}{\sin^2 \theta} \right)$$

$$q_{rot} \sim \int_{-\infty}^{\infty} dp_{\theta} dp_{\phi} \int_{0}^{2\pi} d\phi \int_{0}^{\pi} d\theta e^{-\beta H} = 8\pi^{2} IkT$$

$$q_{vib} \sim \int_{-\infty}^{\infty} dp \int_{-\infty}^{\infty} dx e^{-\beta H} = \frac{kT}{\upsilon}$$

$$\upsilon = \frac{1}{2\pi} \sqrt{\frac{k}{\mu}}$$

transl. missing 
$$\frac{1}{h^3}$$
; rot. missing  $\frac{1}{h^2}$ ; vib. missing  $\frac{1}{h}$ 

Use 
$$q = \sum_{j} e^{-\beta \varepsilon_{j}} \rightarrow \frac{1}{h^{s}} \int \cdots \int e^{-\beta H} \prod_{j=1}^{s} dp_{j} dq_{j}$$
 Add appropriate  $\frac{1}{h^{s}}$  correction factor to the

At high 
$$Q = \frac{q^N}{N!} = \frac{1}{N!h^{sN}} \int \cdots \int e^{-\beta H} \prod_{i=1}^{sN} dp_i dq_i$$
 enough T

correction factor to the classical partition function

We speculate that

$$Q = \frac{1}{N!h^{sN}} \int \cdots \int e^{-\beta H} dp dq$$
Hamiltonian for a system of the system of t

H is the classical n-body Hamiltonian for a system of

products of all the  $dp_i$ ,  $dq_i$ 

Assume that one has a monotonic gas

$$H = \frac{1}{2m} \sum (p_{xj}^2 + p_{yj}^2 + p_{zj}^2) + U(x_1, y_1, z_1, ..., x_N, y_N, z_N)$$

integrate over momenta

We already know that it is a poor approximation to treat vibration classically

Divide H into classical and quantum parts

$$H = H_{cl} + H_{quant}$$
 
$$q = q_{cl} q_{quant}$$

$$Q = Q_{cl}Q_{quant}$$

$$= \frac{Q_{quant}}{N!h^{sN}} \int e^{-H_{cl}/kT} dp_{cl} dq_{cl}$$

 $= \frac{Q_{quant}}{N!h^{sN}} \int e^{-H_{cl}/kT} dp_{cl} dq_{cl}$  s now refers to the number of degrees of freedom treated classically

$$\overline{\varepsilon} = \frac{\int \int He^{-\beta H} dp_1...dq_s}{\int \int e^{-\beta H} dp_r...dq_s} \quad \longleftarrow$$

avg. energy for a molecule in a system of independent molecules

In the special case that

$$H = \sum_{j}^{m} a_{j} p_{j}^{2} + \sum_{j}^{n} b_{j} q_{j}^{2} + H\left(p_{m+1}, ... p_{s}, q_{n+1} ... q_{s}\right)$$

$$kT/2 \qquad kT/2$$
for each degree
of freedom
$$Equipartition of energy$$

transl. = 
$$3/2kT$$
  
rigid rotor =  $kT$   
harmonic osc. =  $kT$   
valid only at very  
high T

$$C_{\rm v} = \frac{5}{2}Nk + Nk \frac{\frac{\theta_{\rm v}^2}{T^2}e^{\theta_{\rm v}/T}}{\left(e^{\theta_{\rm v}/T} - 1\right)^2}$$
 trans