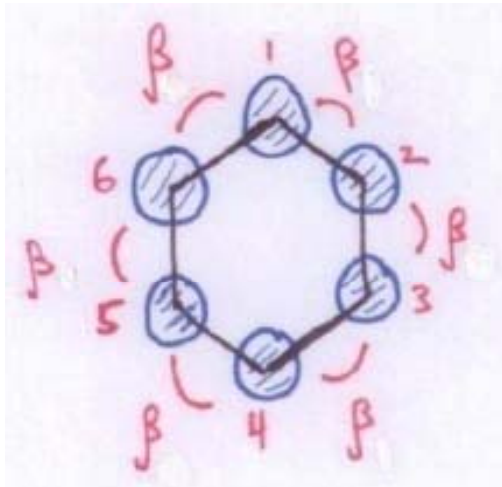


Hückel Theory of Pi-Bonding in Conjugated Hydrocarbon Molecules (Pt. 2)

1) Revisit the Hückel model of pi-bonding in a fully symmetric hydrocarbon:



Benzene, N=6

Consider the matrix Schrodinger Eq. (details on next slide!): $\mathbf{H}\vec{c}^{(j)} = E_j \vec{c}^{(j)}$

Proposed j'th (unnormalized) eigenvector (j=1-6) for benzene (N=6):

$$\mathbf{H} \begin{bmatrix} 0 & \beta & 0 & 0 & 0 & \beta \\ \beta & 0 & \beta & 0 & 0 & 0 \\ 0 & \beta & 0 & \beta & 0 & 0 \\ 0 & 0 & \beta & 0 & \beta & 0 \\ 0 & 0 & 0 & \beta & 0 & \beta \\ \beta & 0 & 0 & 0 & \beta & 0 \end{bmatrix} \begin{pmatrix} e^{2\pi i j \cdot 1/N} \\ e^{2\pi i j \cdot 2/N} \\ e^{2\pi i j \cdot 3/N} \\ e^{2\pi i j \cdot 4/N} \\ e^{2\pi i j \cdot 5/N} \\ e^{2\pi i j \cdot 6/N} \end{pmatrix} = E_j' \begin{pmatrix} e^{2\pi i j \cdot 1/N} \\ e^{2\pi i j \cdot 2/N} \\ e^{2\pi i j \cdot 3/N} \\ e^{2\pi i j \cdot 4/N} \\ e^{2\pi i j \cdot 5/N} \\ e^{2\pi i j \cdot 6/N} \end{pmatrix} ; E_j' = E_j - \alpha$$

$\vec{c}^{(j)}$

N.B.: $i = \sqrt{-1}$ here!

Consider row 3: $\beta(e^{2\pi i j \cdot 2/N} + e^{2\pi i j \cdot 4/N}) = E_j' e^{2\pi i j \cdot 3/N}$

Or: $2\beta \cos(2\pi j / N) = E_j'$

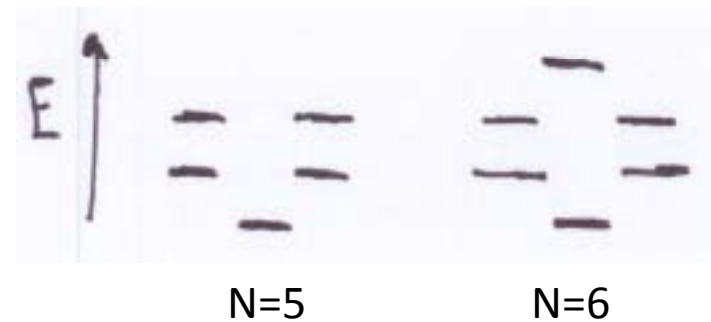
Importantly, the same condition applies for all N=6 rows!

Thus, the following are eigenvector/eigenvalue pairs of \mathbf{H} :

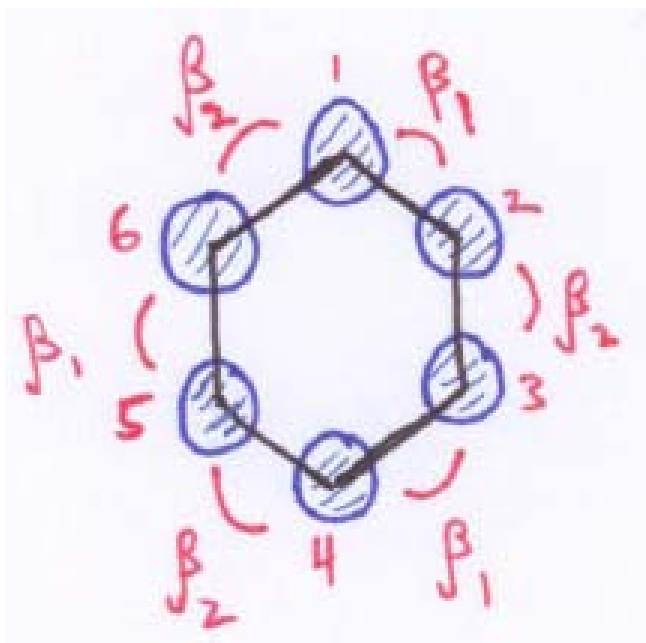
$$\vec{c}^{(j)} \leftrightarrow E_j' = E_j - \alpha = 2\beta \cos(2\pi j / N) \quad \text{for} \quad j = 1, 2, \dots, N$$

- Notes:**
- 1) Same eigenvalue spectrum quoted in previous notes (good!)
 - 2) To normalize the (unnormalized) version of $c^{(j)}$ above, multiply $\times 1/\sqrt{N}$.
 - 3) There is degeneracy in the energy spectrum (see [figure below](#)).
 - 4) Range of j : any continuous patch of j 's will do.

Degeneracy patterns for pi-bonding
MO energies in even/odd cyclic hydrocarbons:



II) Hückel model of pi-bonding in an “alternating hydrocarbon” chain, i.e., one with two alternating values of the Hückel coupling parameter β .



Example, $N=6$

Proposed j'th eigenvector:



$$\begin{bmatrix} 0 & \beta_1 & 0 & 0 & 0 & \beta_2 \\ \beta_1 & 0 & \beta_2 & 0 & 0 & 0 \\ 0 & \beta_2 & 0 & \beta_1 & 0 & 0 \\ 0 & 0 & \beta_1 & 0 & \beta_2 & 0 \\ 0 & 0 & 0 & \beta_2 & 0 & \beta_1 \\ \beta_2 & 0 & 0 & 0 & \beta_1 & 0 \end{bmatrix} \begin{pmatrix} ae^{2\pi ij \cdot 1/N} \\ be^{2\pi ij \cdot 2/N} \\ ae^{2\pi ij \cdot 3/N} \\ be^{2\pi ij \cdot 4/N} \\ ae^{2\pi ij \cdot 5/N} \\ be^{2\pi ij \cdot 6/N} \end{pmatrix} = E' \begin{pmatrix} ae^{2\pi ij \cdot 1/N} \\ be^{2\pi ij \cdot 2/N} \\ ae^{2\pi ij \cdot 3/N} \\ be^{2\pi ij \cdot 4/N} \\ ae^{2\pi ij \cdot 5/N} \\ be^{2\pi ij \cdot 6/N} \end{pmatrix} ; E' = E - \alpha$$

Picking any two sequential “(a,b)” rows in the above Schrodinger matrix equation, leads to a 2-dim. eigenvalue/eigenvector problem that determines a,b and E’.

For example, focusing on rows 3,4:

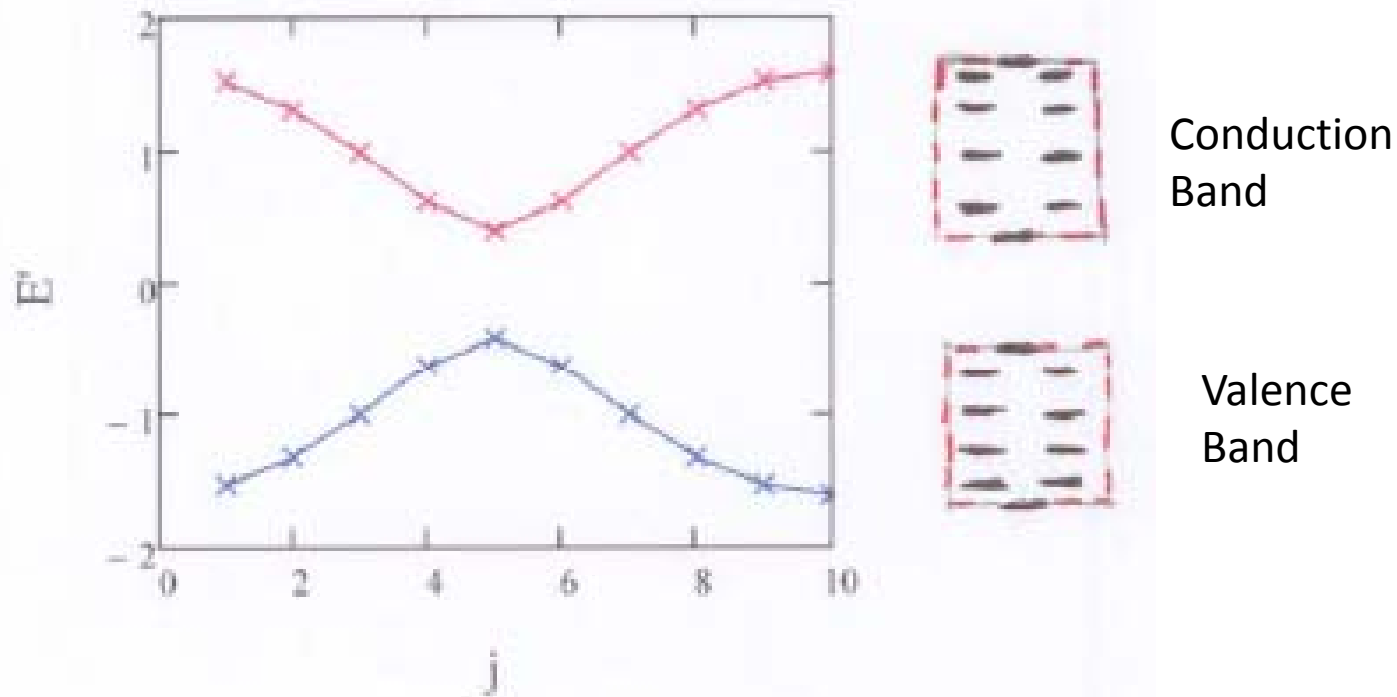
$$\begin{bmatrix} 0 & \beta_2 e^{-2\pi ij/N} + \beta_1 e^{2\pi ij/N} \\ \beta_2 e^{2\pi ij/N} + \beta_1 e^{-2\pi ij/N} & 0 \end{bmatrix} \begin{pmatrix} a \\ b \end{pmatrix} = E' \begin{pmatrix} a \\ b \end{pmatrix}$$

Solving this 2-dim. eigenvalue/eigenvector yields:

$$E_j' = \pm \left[\beta_1^2 + \beta_2^2 + 2\beta_1\beta_2 \cos(4\pi j / N) \right]^{1/2}$$

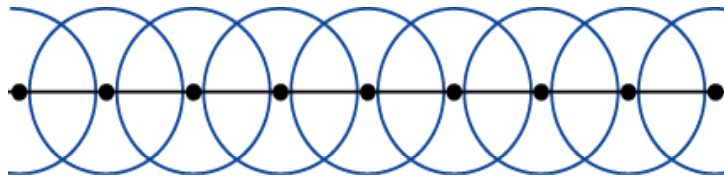
Note: Importantly, for this 2-dim eigenvalue/vector problem to result in the cases of rows (1,2) and (N-1,N), must have $j = \text{integer}$.

Both +/- branches of E' are plotted here for $N=20$ and $\beta_1 = -0.6$, $\beta_2 = -1.0$

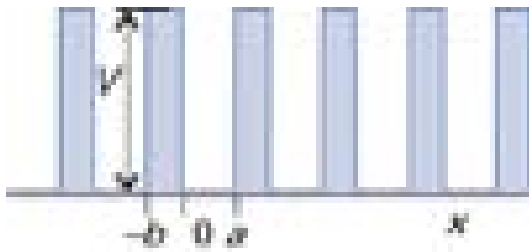


Further points:

- 1) Since there are a total of N unique eigenvectors for an $N \times N$ matrix, the range of j should be restricted to $j=1,2,\dots,N/2$.
- 2) If there are N electrons contributed to the π system (one from each $2p_z$ orbital), then, using the Aufbau principle (2 electrons per MO), the ground electronic state configuration of the alternating hydrocarbon system defined above is: the entire valence band is filled while the entire conduction band is empty.
- 3) Similar band gaps occur for an electron moving in any periodic potential. Such potentials naturally occur in solids:



Periodic array of atoms (and “attached” orbitals) in a 1D solid.



Periodic potential $V(x)$ that the free electrons in this solid move in.