

Chem 2430 Answers HW #1

1. $\psi = e^{-ax^2}$. We want to find N such that $N \int_{-\infty}^{\infty} e^{-2ax^2} dx = 1$

$$N^2 \sqrt{\frac{\pi}{2a}} = 1 \quad \text{So } N = \left(\frac{2a}{\pi}\right)^{1/4}$$

What is the probability of finding the particle between 0 and 1?

$$\text{Prob} = \left(\frac{2a}{\pi}\right)^{1/2} \int_0^1 e^{-2ax^2} dx = \frac{\text{erf}(\sqrt{2a})}{2}$$

2. Show that the $n=2$ and $n=4$ particle-in-a-box wavefunctions are orthogonal.

$$\begin{aligned} \int_0^L \sin\left(\frac{2\pi x}{L}\right) \sin\left(\frac{4\pi x}{L}\right) dx &= a \int_0^L \left[\sin\left(\frac{2\pi x}{L}\right)\right]^2 \cos\left(\frac{2\pi x}{L}\right) dx \\ &= \frac{L}{3\pi} \sin^3\left(\frac{2\pi x}{L}\right) \Big|_0^L = 0 \end{aligned}$$

$$3. \psi = c_1 e^{ikx} + c_2 e^{-ikx}$$

$$\psi(0) = 0 \Rightarrow c_2 = -c_1$$

$$\psi = c_1 (e^{ikx} - e^{-ikx}) = c \sin(kx)$$

$$E = \frac{\hbar^2 k^2}{2m}$$

All positive values of E are possible