

Chem 2430 Exam # 1

$$(1) \psi = \frac{1}{\sqrt{3}} \psi_1 + \sqrt{\frac{2}{3}} \psi_2 \Rightarrow (a) E = \frac{1}{3} E_1 + \frac{2}{3} E_2 = \frac{3}{2} \frac{\hbar^2 \pi^2}{2mL^2}$$

$$(b) \langle \psi | x | \psi \rangle = \frac{1}{3} \langle \psi_1 | x | \psi_1 \rangle + \frac{2}{3} \langle \psi_2 | x | \psi_2 \rangle + \frac{2\sqrt{2}}{3} \langle \psi_1 | x | \psi_2 \rangle$$

$$= \frac{1}{3} \frac{L}{2} + \frac{2}{3} \frac{L}{2} + \frac{2\sqrt{2}}{3} \frac{2}{L} \int_0^L \sin\left(\frac{\pi x}{L}\right) x \sin\left(\frac{2\pi x}{L}\right) dx$$

$$= 0.50L - \frac{32}{27} \frac{\sqrt{2} L}{\pi^2} = 0.50L - 0.17L = 0.33L$$

(2) This is a stationary state as it is an eigenfunction of H .

$$(3) (a) \langle 0 | (a^\dagger + a)(a^\dagger - a) | 0 \rangle = \langle 0 | a a^\dagger | 0 \rangle = 1$$

$$(b) \langle 1 | (a^\dagger - a)(a^\dagger + a) | 1 \rangle = \langle 1 | a^\dagger a - a a^\dagger | 1 \rangle = 1 - 2 = -1$$

(4) I will stick with atomic units.

$$(a) m_p = 1836 \text{ so } \mu_p = 918 \text{ a.u.}$$

$$\omega = \sqrt{\frac{k}{\mu}} = \sqrt{\frac{0.32}{918}} = 0.0187 \text{ a.u.}$$

$$ZPE = \frac{1}{2} \omega = 0.00935 \text{ a.u.} \quad (2052 \text{ cm}^{-1})$$

(b) To find the classical turning points

$$\frac{1}{2} k x^2 = 0.00935 \Rightarrow x = \pm 0.241 \text{ a.u.}$$

$$\alpha = \sqrt{k\mu} = 17.14$$

$$\text{Prob of being in classical region} = 2\sqrt{\frac{k}{\mu}} \int_0^{0.241} e^{-\alpha x} dx$$

$$= 0.85. \text{ So } 15\% \text{ probability in forbidden region.}$$

(5) Electron on ring to model π electrons of benzene

radius of benzene ~ 3 a.u.

Allow for "size" of atomic orbitals so take $r_0 = 4$ a.u.

$$E = \frac{m^2}{2r_0^2} \text{ (in a.u.)}$$

$$\begin{array}{l} \text{---} \quad \text{---} \quad m = \pm 2 \\ \text{---} \quad \text{---} \quad m = \pm 1 \\ \text{---} \quad \quad m = 0 \end{array}$$

Since benzene has six π electrons, the transition is $m=1 \rightarrow m=2$

$$\text{Excitation energy} = \frac{4-1}{2} \frac{1}{r_0^2} \sim .094 \text{ a.u.} = 2.4 \text{ eV}$$

$$(6) \quad V = \frac{1}{2} k x^2 + \frac{1}{2} k y^2$$

$$(a) \quad E = \hbar \omega (n_x + n_y + 1)$$

$$\text{ZPE} = \hbar \omega$$

The degeneracy grows as $n_x + n_y + 1$

E.g. $(2,0)$, $(1,1)$, and $(0,2)$ are degenerate

$$(b) \quad k_x = k + \delta, \quad k_y = k - \delta$$

$$\begin{aligned} \omega_x &= \sqrt{\frac{k_x}{\mu}} = \sqrt{\frac{k+\delta}{\mu}} = \sqrt{\frac{k}{\mu}} \sqrt{1 + \delta/k} \\ &= \omega \left(1 + \delta/2k - 1/8 \left(\frac{\delta}{k} \right)^2 + \dots \right) \end{aligned}$$

$$\begin{aligned} \omega_y &= \sqrt{\frac{k_y}{\mu}} = \sqrt{\frac{k-\delta}{\mu}} = \sqrt{\frac{k}{\mu}} \sqrt{1 - \delta/k} \\ &= \omega \left(1 - \delta/2k - 1/8 \left(\frac{\delta}{k} \right)^2 - \dots \right) \end{aligned}$$

$$\omega_x + \omega_y = 2\omega - (1/4) \left(\frac{\delta}{k} \right)^2 \omega$$

$$\text{Change in ZPE} = - \frac{\omega}{8} \left(\frac{\delta}{k} \right)^2$$

$$(7) \psi_I = A \sin(kx)$$

$$k = \sqrt{\frac{2mE}{\hbar^2}}$$

$$\psi_{II} = B e^{kx} + C e^{-kx}$$

$$k = \sqrt{\frac{2m(E-V_0)}{\hbar^2}}$$

$$\psi_{III} = D \sin[k(x-2a-b)]$$

$$\psi_I(a) = \psi_{II}(a) \quad \text{and} \quad \psi_I'(a) = \psi_{II}'(a)$$

$$A \sin(ka) = B e^{ka} + C e^{-ka}$$

$$kA \cos(ka) = k(B e^{ka} - C e^{-ka})$$

can get B and C in terms of A and the other quantities.

$$\psi_{II}(a+b) = \psi_{III}(a+b), \quad \psi_{II}'(a+b) = \psi_{III}'(a+b)$$

$$B e^{k(a+b)} + C e^{-k(a+b)} = D \sin(ka)$$

$$k(B e^{k(a+b)} - C e^{-k(a+b)}) = D k \cos(-ka)$$

Can get D in terms of B or C, and thus in terms of A