

The 1D Schrodinger Eq is an example of a second-order linear homogeneous differential eq. (DE)

General solution of a 2<sup>nd</sup> order linear homogeneous DE

$$y'' + P(x)y' + Q(x)y = 0$$

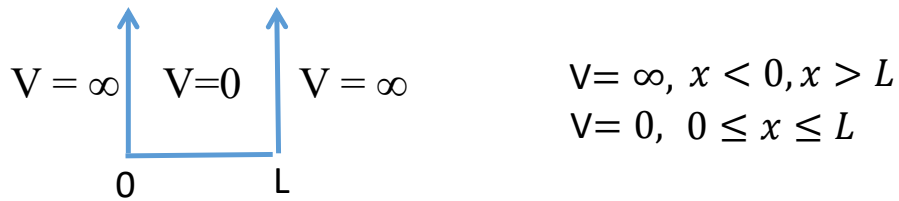
is  $y = c_1 y_1(x) + c_2 y_2(x)$ , where  $y_1$  and  $y_2$  obey the DE

There are two constants consistent with the need to integrate  $y''$  twice to solve the eq.

$$c_1 y_1'' + c_2 y_2'' + P(c_1 y_1' + c_2 y_2') + Q(c_1 y_1 + c_2 y_2) = 0$$

$$c_1 [y_1'' + P y_1' + Q y_1] + c_2 [y_2'' + P y_2' + Q y_2] = 0$$

## The particle in a box problem



For this potential, the particle can only be in the box where  $V=0$

$$-\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} = E\psi \quad \text{inside the box}$$

Possible solutions are:  $\sin(ax)$  and  $\cos(ax)$  where  $a$  is some constant

$$\frac{d^2}{dx^2} \cos(ax) = -a^2 \cos(ax)$$

$$\frac{d^2}{dx^2} \sin(ax) = -a^2 \sin(ax)$$

So a general solution is:  $\psi = c_1 \sin(ax) + c_2 \cos(ax)$

**Now lets apply the boundary conditions**

$$\psi(x) = 0 \text{ at } x = 0, L$$

$-\frac{\hbar^2}{2m} \frac{d^2}{dx^2}$	Is the kinetic energy operator
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Using the boundary condition at  $x = 0$

$$\psi(0) = 0 \implies c_2 = 0, \text{ since } \cos(0) = 1$$

$$\psi = c_1 \sin(ax)$$

Using the boundary condition at  $x = L$

$$\psi(L) = c_1 \sin(aL)$$

$c_1 = 0$  would imply no particle

Nontrivial case:  $c_1 \neq 0$

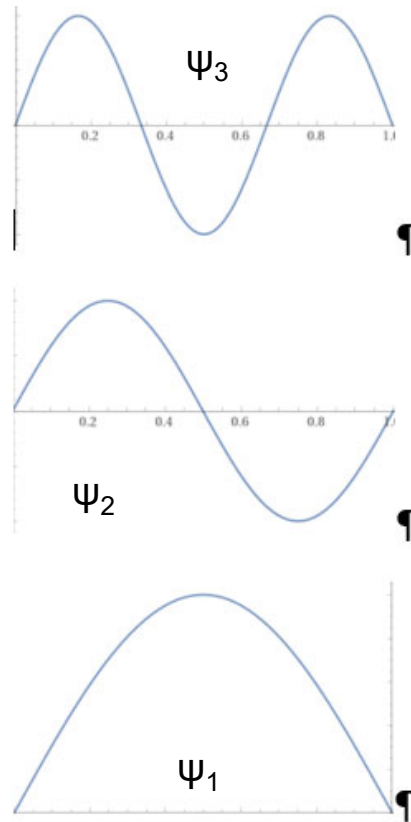
$$\implies aL = n\pi, n = 1, 2, 3, \dots$$

$$\psi(x) = c_1 \sin\left(\frac{n\pi x}{L}\right)$$

$c_1$  is just a normalization constant

Normalize the wave function

$$\psi_n = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$$



Suppose  $\Psi(x,0) = \psi_1 + \psi_2$

$$\Psi(x,t) = a_1 \exp(-iE_1 t/\hbar) \psi_1 + a_2 \exp(-iE_2 t/\hbar) \psi_2$$

For 1D systems, the energy increases with the number of nodes.

Now plug  $\psi$  into SE

$$\frac{\hbar^2}{2m} \left( \frac{n^2 \pi^2}{L^2} \right) = E$$

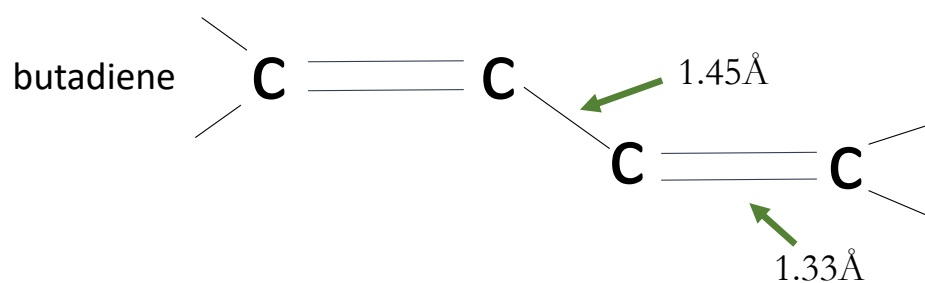
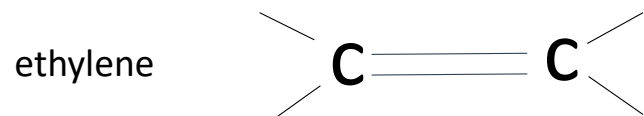
$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2} = \frac{n^2 h^2}{8mL^2}$$

**Things to note:**

- energy levels are quantized
- $E \neq 0$ . Why?
- Levels get closer as  $m \rightarrow \infty$ ,  $L \rightarrow \infty$ , or  $h \rightarrow 0$

I.e., as the system approaches classical behavior

Can use to model  $\pi$  electrons of polyenes



Lets take all bond lengths to be 1.4Å

Since the orbitals extend beyond the C atoms we should add  $\sim 0.5 \text{ \AA}$  to each end

	box size	# $\pi$ electrons
Ethylene	$1.4+1.0=2.4\text{\AA}$	2
Butadiene	$3(1.4)+1.0=5.2\text{\AA}$	4
Hexatriene	$5(1.4)+1.0=8\text{\AA}$	6

Node count

$\psi_1$	0
$\psi_2$	1
$\psi_3$	2



Same # of nodes as the “real”  $\pi$  orbitals of polyenes (not counting the sign change due to reflecting in the plane of the atoms)

From past experience we know that there can be at most 2 electrons per orbital.

What is the energy of the lowest optical absorption?

Ethylene  $\Delta n=1 \rightarrow 2$

Butadiene  $\Delta n=2 \rightarrow 3$

Hexatriene  $\Delta n=3 \rightarrow 4$

## Introduction of atomic units (a.u.)

In atomic units  $\hbar$ ,  $c$ ,  $m_e$  all = 1

Atomic unit of distance	= 1Bohr = 0.529 Å
Atomic unit of energy	= 27.211 eV = 2x the energy of the 1s orbital of the H atom

What would you guess to be the atomic unit of time?

## Butadiene absorption

$$E_n = \frac{n^2 \pi^2 \hbar^2}{2mL^2} = \frac{n^2 \pi^2}{2(10)^2} \sim \frac{n^2 \pi^2}{200} \sim \frac{n^2}{20} = 0.049n^2$$

 Switch to atomic units

n=1          1.33 eV

n=2          5.33 eV

n=3          12.0 eV

Thus  $\Delta E(n=2 \rightarrow 3) = 6.67 \text{ eV}$

red~1.8eV, blue~3.1eV

As the chain length increases, the transition moves from the UV to the visible and then into the infrared.

Recall that the dot product between two vectors is

$$\vec{r} \cdot \vec{s} = r_x s_x + r_y s_y + r_z s_z$$

The dot product is 0 if the vectors are orthogonal

The integral  $\int_{-\infty}^{\infty} \psi_i^* \psi_j dx$  is the continuous function analog of the dot product of vectors in Cartesian space

$$\text{In Dirac notation } \int_{-\infty}^{\infty} \psi_i^* \psi_j dx = \langle \psi_i | \psi_j \rangle$$

For the particle in the box problem

$$\int_0^L \psi_i \psi_j dx = \delta_{ij} = \begin{cases} 0 & \text{if } i \neq j \\ 1 & \text{if } i = j \end{cases}$$

The functions  $\{\psi_i\}$  form an orthonormal (both orthogonal and normalized) set.

The integration runs from 0 to L for the particle in the box problem

They also form a basis set in which any other function on (0,L) can be represented

## Free particle and momentum

Free particle  $\psi = e^{ikx}$  travelling to right

$e^{-ikx}$  travelling to left

momentum operator (for motion in the x direction)

$$\hat{p}_x = \frac{\hbar}{i} \frac{d}{dx}$$

$$p_x e^{ikx} = \hbar k e^{ikx}$$

$$p_x e^{-ikx} = -\hbar k e^{-ikx}$$

For a free particle  $k$  can take on any real value

For a free particle

$$E = \frac{p^2}{2m} = \frac{\hbar^2 k^2}{2m}$$

$A\phi = a\phi$  is an example of an **eigenvalue problem** with  $A$  being some operator,  $a$  being the **eigenvalue**, and  $\phi$  an **eigenfunction**

Note  $\sin(kx)$  and  $\cos(kx)$  are not eigenfunctions of the momentum operator



## Euler's relations

$$e^{ix} = \cos(x) + i \sin(x)$$

$$e^{-ix} = \cos(x) - i \sin(x)$$

$$\cos(x) = (e^{ix} + e^{-ix})/2$$

$$\sin(x) = (e^{ix} - e^{-ix})/2i$$

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Note:  $\exp(i\pi) = -1$