

CHAPTER 12: MEANING OF T

$$\frac{1}{T} = \left(\frac{\partial S}{\partial U} \right)_{V,N}$$

Consider Schottky two-state model

$$\langle \varepsilon \rangle = \frac{\varepsilon_0 e^{-\varepsilon_0/kT}}{1 + e^{-\varepsilon_0/kT}} \quad C_v = \frac{N\varepsilon_0^2}{kT^2} \frac{e^{-\beta\varepsilon_0}}{(1 + e^{-\beta\varepsilon_0})^2}$$

$$\downarrow$$

$$\frac{S}{N} = \frac{\varepsilon_0 e^{-\beta\varepsilon_0}}{T(1 + e^{-\beta\varepsilon_0})} + k \ln(1 + e^{-\beta\varepsilon_0})$$

$$U = n\varepsilon_0 \rightarrow n = \frac{U}{\varepsilon_0}$$

$$W = \frac{N!}{n!(N-n)!}$$

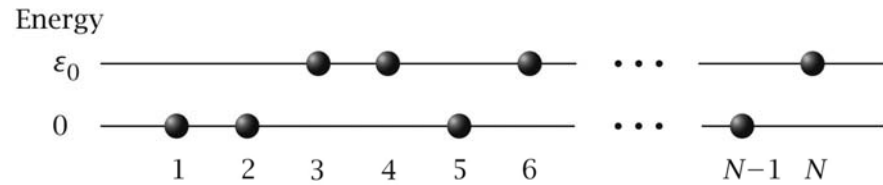
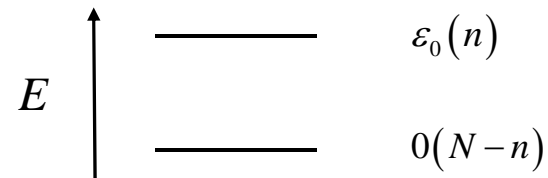


Figure 12.1 Molecular Driving Forces 2/e (© Garland Science 2011)



$$\frac{S}{k} = \ln W = -n \ln \frac{n}{N} - (N-n) \ln \left(\frac{N-n}{N} \right)$$

$$\frac{S}{k} = -\frac{U}{\varepsilon_0} \ln \left(\frac{U}{N\varepsilon_0} \right) - \left(N - \frac{U}{\varepsilon_0} \right) \ln \left(\frac{N - U/\varepsilon_0}{N} \right)$$

$$\begin{aligned} \frac{1}{T} &= \left(\frac{\partial S}{\partial U} \right)_{V,N} = \left[\frac{1}{\varepsilon_0} \ln \frac{U}{N\varepsilon_0} - \frac{U}{\varepsilon_0} \frac{1}{U} + \frac{1}{\varepsilon_0} \ln N \left(\frac{N - U/\varepsilon_0}{N} \right) - \left(N - \frac{U}{\varepsilon_0} \right) \frac{-1/\varepsilon_0}{N - U/\varepsilon_0} \right] k \\ &= \left[\frac{1}{\varepsilon_0} \ln \frac{n}{N} - \frac{1}{\varepsilon_0} + \frac{1}{\varepsilon_0} \left(\frac{N-n}{N} \right) - \frac{(N-n)(-1/\varepsilon_0)}{N-n} \right] k \\ &= +\frac{1}{\varepsilon_0} \left[-1 - \ln \frac{n}{N} + \ln \frac{N-n}{N} + 1 \right] k \\ &= \frac{1}{\varepsilon_0} \left[\ln \left(\frac{N-n}{N} \right) - \ln \left(\frac{n}{N} \right) \right] k \\ &= \frac{-k}{\varepsilon_0} \ln \left(\frac{n/N}{(N-n/N)} \right) = \frac{k}{\varepsilon_0} \left(\frac{f_{ground}}{f_{excited}} \right) \quad \left| \quad f = \text{fraction} \right. \end{aligned}$$

so T depends on ε_0 , N , U

spacing of levels
particles
total energy

Now consider 3 particles in 2 levels

low $U, \rightarrow f_{ground} / f_{exc} > 1$, so T positive

absorb energy to $> S$

$f_{ground} = f_{excited}, 1/T = 0$ ($T = \infty$)
if

$S(U)$ is at maximum

equal populations $\rightarrow T = \infty$

if more particles in
upper level T is negative

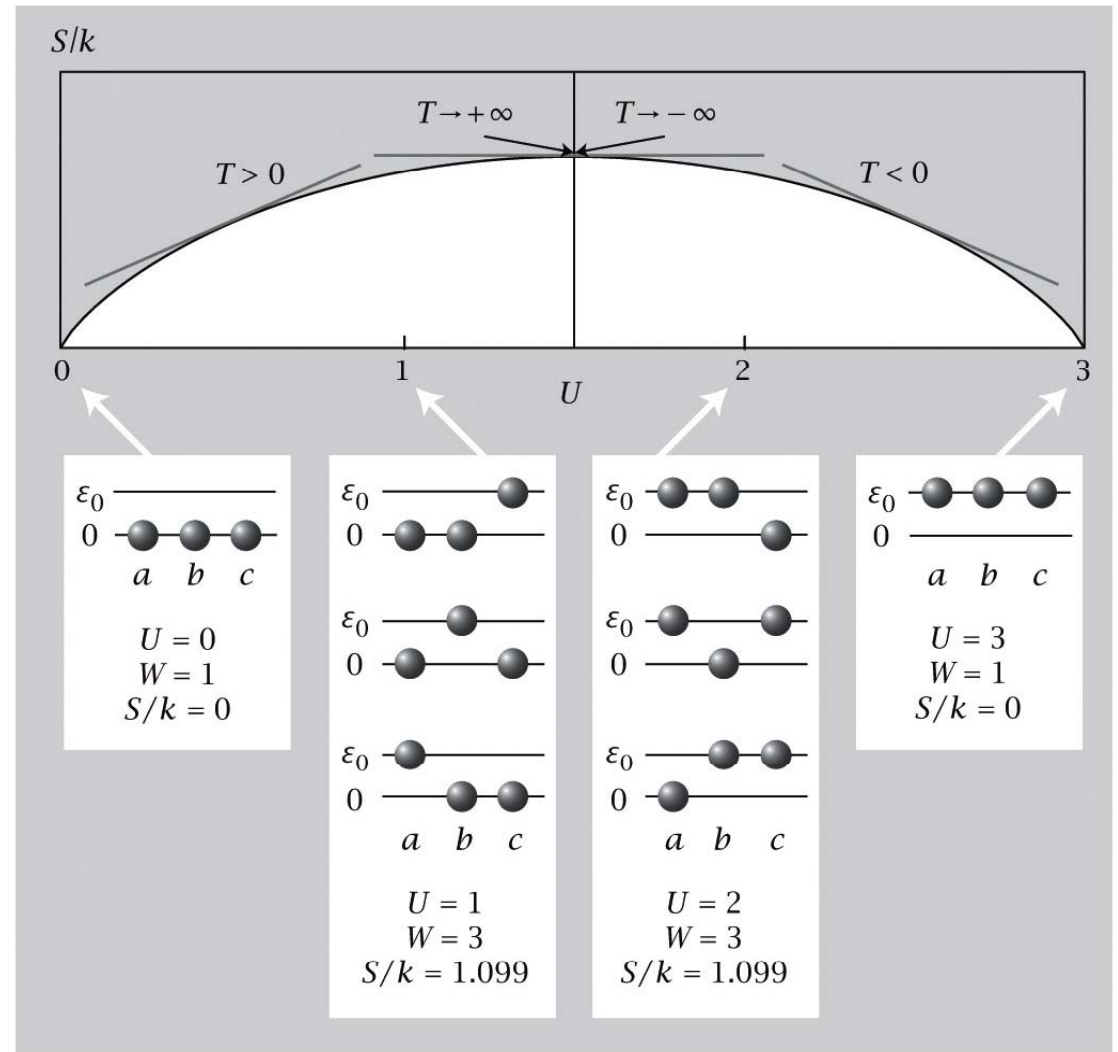


Figure 12.2 Molecular Driving Forces 2/e (© Garland Science 2011)

A system at $-T$ is hotter than at $+T$

This system would tend to give off energy,
not to absorb it.

$\frac{1}{T} > 0$, system has tendency to absorb energy

$\frac{1}{T} < 0$, system has tendency to lose energy

Negative T only happens in systems with
finite # levels and are saturable.

negative T systems \leftrightarrow population inversion

cannot be achieved by equilibrium with
heat bath, since heat bath has $+T$

$$dS = \frac{\delta q}{T}: \text{Reflection on this result}$$

degrees of freedom U, V, N , hold V, N constant, so work = 0

$$\Rightarrow dU = \delta q$$
$$dS = \frac{dU}{T} = \frac{\delta q}{T} \quad \text{system acquires heat from heat bath, due to increasing excited state population}$$

Ideal gas, *const V* $S = \frac{3Nk}{2} \ln U$

$$S = 3/2 Nk \ln(U) \quad \text{fixed } V, N$$

$$\frac{1}{T} = \left(\frac{\partial S}{\partial U} \right)_{V,N} \quad U = 3/2 NkT \quad T = 2/3 U / Nk$$

$$dS = \frac{dU}{T} \quad dS = \frac{Nk dU}{2/3 U} = 3/2 Nk \frac{dU}{U}$$

$$\frac{1}{T_A} = \frac{1}{T_B} \Rightarrow \frac{1}{\epsilon_A} \ln \left(\frac{N_A \epsilon_A}{U_A} \right) = \frac{1}{\epsilon_B} \ln \left(\frac{N_B \epsilon_B}{U_B} \right)$$

same material $\epsilon_A = \epsilon_B$ and same # particles $N_A = N_B$, tendency to maximize S and to equalize temperatures is a tendency to equalize energies

if $\epsilon_A = \epsilon_B$ but $N_A \neq N_B$ then equilibrium when

$$\frac{N_A}{U_A} = \frac{N_B}{U_B}$$

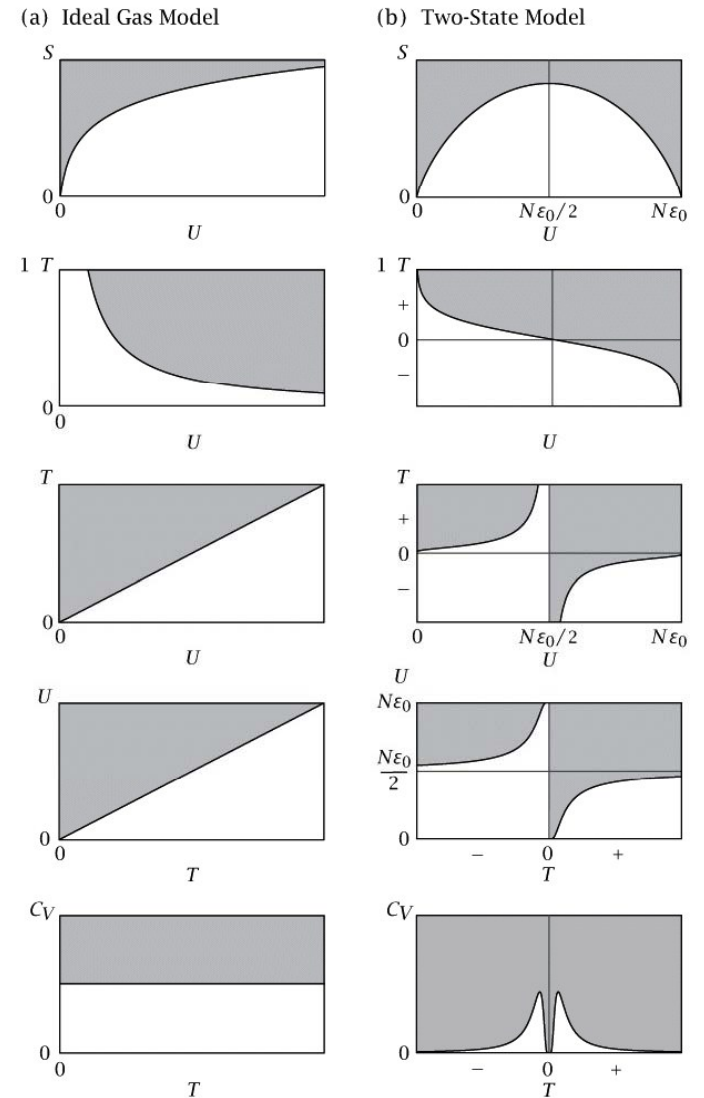


Figure 12.3 Molecular Driving Forces 2/e (© Garland Science 2011)

Heat capacity is a measure of energy fluctuations

$$p(E) = \frac{1}{2} W(E) e^{-\beta E} \quad \text{Highly peaked} \\ \text{width determined by } C_V$$

$$\ln p(E) = \ln p(U) + \left(\frac{\partial \ln p(E)}{\partial E} \right)_{E=U} (E-U) + \frac{1}{2} \left(\frac{\partial^2 \ln p(E)}{\partial E^2} \right)_{E=U} (E-U)^2$$

at peak $S(E) = S(U)$, equil. value

$$\frac{\partial \ln p(E)}{\partial E} = \frac{1}{kT(E)} - \frac{1}{kT_0}$$

$$\frac{\partial^2 \ln(p(E))}{\partial E^2} = \frac{-1}{kT^2} \left(\frac{\partial T}{\partial E} \right)_{E=U} = - \left(\frac{1}{kT_0^2 C_V} \right)$$

$$p(E) = p(U) e^{\left[\frac{-(E-U)^2}{2kT^2 C_V} \right]} = e^{[U - T_0 S(U)]} e^{-\frac{(E-U)^2}{2kT^2 C_V}}$$

$$\Rightarrow \sigma^2 = \langle (E-U)^2 \rangle = \langle E^2 \rangle - U^2 = kT_0^2 C_V$$

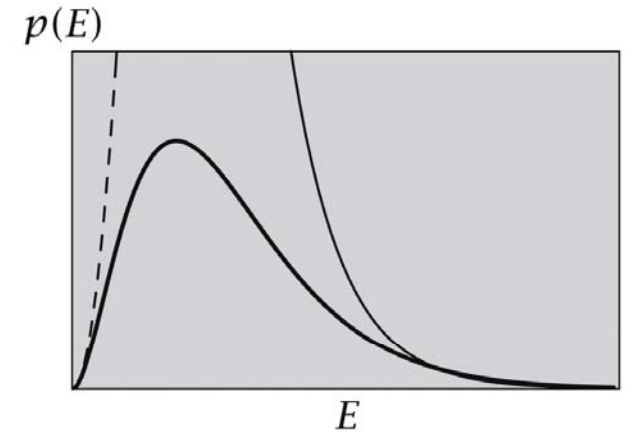


Figure 12.5 Molecular Driving Forces 2/e (© Garland Science 2011)

Ideal gas

$$\frac{\sigma}{U} = \frac{\sqrt{kT^2 C_v}}{3/2 NkT} = \left(\frac{3}{2} N\right)^{-1/2}$$

$$N \sim 10^{23} \Rightarrow \frac{\sigma}{U} \sim 10^{-12}$$

Relatively speaking, the fluctuations are a very small amount of the total energy when N is large