

Answers: HW #4, Chem 1410

1. radial part of  $\psi_{2p_z}$  (in a.u.) =  $r e^{-r/2}$   
 (Note that if we just want to find the maximum, we don't need to normalize)

Radial  $\propto r^2 \psi^2 = r^4 e^{-r}$        $\frac{dP}{dr} = (4r^3 - r^4) e^{-r}$   
 $\Rightarrow r_{\max} = 4 \text{ a.u.}$

2.  $\psi = \frac{1}{\sqrt{2}}(\psi_{2s} + \psi_{2p_z})$ : What is the probability of finding the electron between  $\theta=0$  and  $\theta=\pi/2$ . By symmetry it is  $1/2$ . Do you see this?

$$\psi^2 = \frac{1}{2} \frac{1}{32\pi} ((2-r)^2 + r^2 \cos^2 \theta + 2r(2-r)\cos \theta) e^{-r}$$

$$\int_0^\infty r^2 (2-r)^2 e^{-r} dr = 8, \quad \int_0^\infty r^4 e^{-r} dr = 24, \quad \int_0^\infty \frac{1}{2} r^3 (2-r) e^{-r} dr = -24$$

$$\int_0^{\pi/2} \sin^2 \theta d\theta = 1, \quad \int_0^{\pi/2} \cos \theta \sin \theta d\theta = 1/2, \quad \int_0^{\pi/2} \cos^2 \theta \sin \theta d\theta = 1/3$$

$$\frac{2\pi}{64\pi} \int \psi^2 r^2 \sin \theta dr d\theta = \frac{1}{32} [8 + 24/3 - 24/2] = \frac{1}{32} [8 + 8 - 12] = 1/8$$

3.  $\left[ -\frac{1}{2} \frac{d^2}{dr^2} - \frac{1}{r} \frac{d}{dr} - \frac{1}{r} + \frac{\hat{L}^2}{2r^2} \right] \psi = E \psi$

$\psi = r e^{-r/2}$  | radial part only.  $\hat{L}^2$  on angular part gives  $1(z)$  since this is a p orbital

$$\psi' = (1 - \frac{r}{2}) e^{-r/2}; \quad \psi'' = (-\frac{1}{2} - \frac{1}{2} + \frac{r}{4}) e^{-r/2} = (-1 + \frac{r}{4}) e^{-r/2}$$

$$-\frac{1}{2} \left[ -1 + \frac{r}{4} \right] - \frac{1}{r} \left( 1 - \frac{r}{2} \right) - \frac{r}{r} + \frac{2}{2r} = E r$$

$$\Rightarrow \left( -\frac{1}{2} + \frac{1}{2} \right) + \left( -\frac{r}{8} - E r \right) + \left( -\frac{1}{r} + \frac{1}{r} \right) = 0, \quad \Rightarrow E = -\frac{1}{8} \text{ a.u.}$$

4.  $\frac{4\pi}{\pi} \int_0^{1.9 \times 10^{-5}} r^2 e^{-2r} dr = 4(228 \times 10^{-15}) \text{ a.u.}$